

Monobromoisophakellin, a New Bromopyrrole Alkaloid from the Caribbean Sponge *Agelas* sp.

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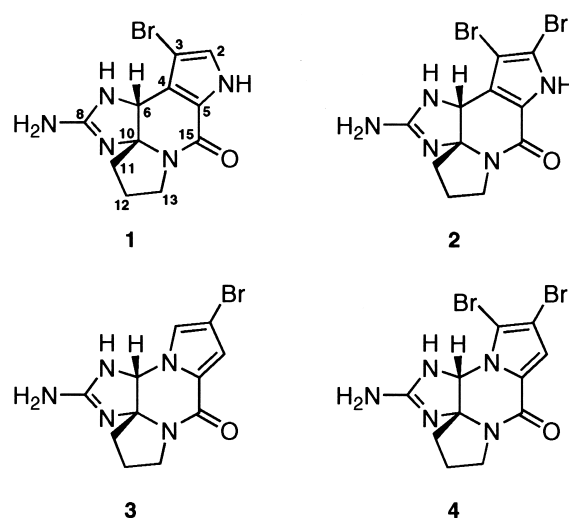
A detailed analysis of the chemical constituents of a Caribbean specimen of *Agelas* sp. was carried out. Four brominated compounds (**1**–**4**) were isolated and one of them was identified as a new bromopyrrole metabolite, monobromoisophakellin (**1**). The structure of **1** was determined using spectroscopic methods. All compounds were tested for their antifeedant activity against the Caribbean reef fish *Thalassoma bifasciatum* in an aquarium assay.

Introduction

Bromopyrrole alkaloids are well known in marine sponges of the genus *Agelas* (Braekman *et al.*, 1992). In our search for bioactive substances from marine organisms, a series of brominated pyrrole alkaloids have been isolated from a specimen of the Caribbean sponge *Agelas* sp. collected off the coast of Sweetings Cay (Bahamas). Examination of the dichloromethane/methanol extract of this sponge resulted in isolation of the known alkaloids dibromoisophakellin (**2**), which has been previously isolated from *Acanthella carteri* (Fedoreyev *et al.*, 1986), monobromophakellin (**3**), and dibromophakellin (**4**), both previously described from *Phakellia flabellata* (Sharma and Burkholder, 1971; Sharma and Magdoff-Fairchild, 1977) as well as of the new bromopyrrole alkaloid, monobromoisophakellin (**1**). In this communication we describe the isolation and structural elucidation of the new bromopyrrole alkaloid (**1**).

Materials and Methods

The marine sponge *Agelas* sp. employed in this study was collected in September 1998 by SCUBA diving (15 m depth) off the coast of Sweetings Cay (Grand Bahama Island) during a scientific cruise of the *R/V Edwin Link* to the Bahamas. The specimen is an undescribed species of *Agelas* (order Agelasida, family Agelasidae), the colour in life is reddish-orange, consistency is tough, spongy, firm



Scheme 1

and almost incompressible. A voucher fragment has been deposited under registration no. ZMA POR. 13369 in the Zoölogisch Museum, Amsterdam, The Netherlands.

Samples of *Agelas* sp. were immediately frozen after collection and kept at -20°C until extraction. For bulk extraction followed by isolation of brominated secondary metabolites, lyophilized sponge tissue (102 g) was ground and extracted exhaustively in a 1:1-mixture of dichloromethane/MeOH at room temperature. The orange-colored wet crude extract was partitioned between *n*-hex-

ane (3×500 ml) and MeOH (150 ml). The obtained MeOH extract was finally partitioned between *n*-BuOH (3×500 ml) and H₂O (300 ml). The resulting *n*-BuOH phase (5.9 g) was purified by gel permeation chromatography on LH-20 Sephadex (Pharmacia) using MeOH as mobile phase. Final purification of the isolated compounds was achieved by preparative RP₁₈ HPLC (details see figure caption 1) to afford **1** (86 mg, 0.08% of dry weight), **2** (28 mg, 0.03%), **3** (35 mg, 0.03%) and **4** (148 mg, 0.15%).

¹H NMR and ¹³C NMR spectra were recorded on a Bruker DRX600 NMR spectrometer. All NMR experiments were measured at 300 K. The 2D experiments (¹H,¹H-COSY, ¹H,¹³C-HSQC, ¹H,¹³C-HMBC, ¹H,¹⁵N-HSQC and ¹H,¹⁵N-HMBC) were carried out using standard parameters. Mass spectral analysis (HRFABMS) was performed on a JEOL JMS-700 sector-field mass spectrometer with 3-nitrobenzyl alcohol (NBA) as matrix or using a Fison VG Platform II for ESIMS. HPLC analysis was carried out as previously reported (Assmann *et al.*, 1999; Assmann *et al.*, 2000). IR (KBr) spectra were recorded on a Perkin Elmer 1600 Series FT-IR spectrometer. UV/VIS spectra were obtained using a Perkin Elmer UV/VIS spectrometer Lambda 16.

Bromopyrrole alkaloids are known to be the principal defensive strategy of Caribbean sponges against predatory reef fishes (Pawlik *et al.*, 1995; Chanas *et al.*, 1996; Wilson *et al.*, 1999; Assmann *et al.*, 2000; Assmann *et al.*, 2001). To investigate the antifeedant activity of the four metabolites, aquarium assays were performed using previously described methods (Pawlik *et al.*, 1987; Pawlik *et al.*, 1995; Chanas *et al.*, 1996; Assmann *et al.*, 2000).

Results and Discussion

The compounds **1–4** could be isolated from the *n*-BuOH phase of *Agelas* species. The brominated alkaloids dibromoisophakellin (**2**), monobromophakellin (**3**), and dibromophakellin (**4**) were identified by comparison of their spectroscopic data with those previously reported (Sharma and Burkholder, 1971 → **4**; Sharma and Magdoff-Fairchild, 1977 → **3** + **4**; De Nanteuil *et al.*, 1985 → **4**; Fedoreyev *et al.*, 1986 → **2**; Jiménez and Crews, 1994 → **4**). The ESI mass spectrum (negative ion mode) of the new compound monobromoisophakellin (**1**) showed prominent pseudo-

Table I. ¹H, ¹³C and ¹⁵N NMR chemical shifts (δ) of **1** in DMSO-*d*₆.

Position		δ(¹³ C)/δ(¹⁵ N) ^a	δ(¹ H) ^b
1	NH	155	12.44 (1H, br)
2	CH	124.4	7.22 (1H, d), <i>J</i> = 3.0 Hz
3	C	93.3	–
4	C	121.6	–
5	C	121.4	–
6	CH	54.1	5.23 (1H, s)
7	NH	89	8.88 (1H, br)
8	C	157.0	–
9	NH	109	9.96 (1H, br)
10	C	84.2	–
11	CH ₂	39.1	2.22 (2H, m)
12	CH ₂	19.2	2.04 (2H, m)
13	CH ₂	43.9	3.57/3.47 (2H, m)
14	N	123	–
15	C	155.5	–
16	NH ₂	72	8.07 (2H, br)

^a ¹³C chemical shifts are given in [ppm] and are referenced to the DMSO-*d*₆ signal (39.5 ppm). ¹⁵N chemical shifts are given in [ppm] and are calibrated according to the Bruker frequency, which is set to 0 ppm for NH₃, the accuracy is about 1 to 2 ppm.

^b ¹H chemical shifts are given in [ppm] and are referenced to the DMSO-*d*₆ signal (2.50 ppm). The integration and the multiplicity of the proton signals are given in parenthesis.

molecular ion peaks at *m/z* 308 and 310 in the ratio 1:1, suggesting the presence of one bromine atom. The molecular formula of **1** was established as C₁₁H₁₃BrN₄O by HRFABMS (*m/z* 310.0290, [M + H]⁺, Δ –1.3 mmu), which is in accordance with the ¹H and ¹³C NMR data (see Table I). The presence of a pyrrole ring conjugated with a carbonyl group part was supported by the UV absorption (MeOH) at λ_{max} 276 nm (lg ε 3.84 mol^{–1}cm^{–1}). The signal at δ_C 155.5 ppm was attributed to a carbonyl group which further supported by the IR (KBr) absorption band at ν_{max} 1697 cm^{–1}. The signal at δ_C 157.0 ppm is typical for the bromopyrrole alkaloids and is assigned to the carbon atom of the guanidine (C-8). From the ¹H,¹³C-HMBC spectrum 28 correlations and from the ¹H,¹⁵N-HMBC spectrum 7 correlations could be obtained which confirmed the proposed structure of **1**. Due to the ¹⁵N data it was possible to distinguish between N-7 and N-9 (the aminoimidazol is protonated). The absolute configuration of **1** was obtained by comparison of the CD spectral data (*c* = 82 μol/l, MeOH, [θ]₂₁₀ – 1220) with the values published in the literature (Fedoreyev *et al.*, 1986).

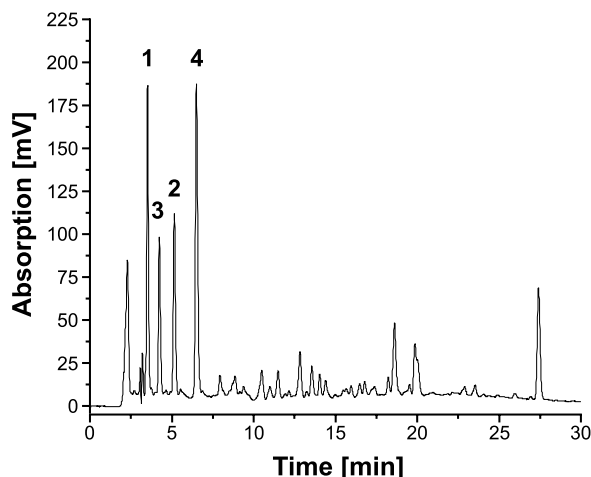


Fig. 1. HPLC profile of DCM/MeOH crude extract from *Agelas* sp. (column: Kromasil RP18, 4×250 mm, $5 \mu\text{m}$; gradient: 20–50% MeCN/ H_2O + 0.1% TFA in 30 min; flow rate: 1 ml/min, UV detection at 280 nm). The retention times for *Agelas* sp. are: monobromoisophakellin (**1**) $t_R = 3.52$ min, monobromophakellin (**3**) $t_R = 4.23$ min, dibromoisophakellin (**2**) $t_R = 5.15$ min, and dibromophakellin (**4**) $t_R = 6.50$ min.

The results of the antifeedant activity of the four phakellin derivatives against *Thalassoma bifasciatum* in the aquarium assay are given in Table II. This shows a higher activity for the isophakelline skeleton in comparison to the phakelline skeleton. The isophakellins which are active at 1 mg/ml are in the same range of the antifeedant activity as oroidin (Chanas *et al.*, 1996; Assmann *et al.*, 2000). It is further known from the literature that bromination increases the antifeedant activity which is confirmed by presented results (Assmann *et al.*, 2000). In contrast to other brominated alkaloids the natural concentration of the four compounds in *Agelas* sp. is relatively low (see Table II). All concentrations (0.05–0.26 mg/ml) are below the required concentration for feeding detergency (1 mg/ml for **1** and **2** and even higher for **3** and **4**).

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Table II. Results of the aquarium assays for compounds **1** to **4** at different concentrations^a.

Compound	1 mg/ml ^b	5 mg/ml ^b	10 mg/ml ^b	Nat. conc. ^c
1	6.0 ± 1.0	3.3 ± 0.6	0.7 ± 0.6	0.15 mg/ml
2	4.3 ± 0.6	1.7 ± 0.6	0	0.05 mg/ml
3	8.7 ± 0.6	6.7 ± 0.6	2.7 ± 0.6	0.06 mg/ml
4	7.7 ± 0.6	5.0 ± 1.0	1.3 ± 0.6	0.26 mg/ml

^a Aquarium assay results of feeding by *Thalassoma bifasciatum* on pellets treated with purified bromopyrrole alkaloids (**1** to **4**) isolated from *Agelas* species. All control pellets were eaten in all assays. Three replicate assays were performed at each concentration (mean \pm SD are indicated in columns 2 to 4). For any individual assay, a treatment was considered deterrent if the number of pellets eaten was less than or equal to 6 ($p < 0.043$, Fisher exact test, 1-tailed; Zar, 1999).

^b Concentration of the pure compound (**1** to **4**) in the pellet. The molar concentrations are $3.22 \mu\text{M}$ (**1** and **3**) and $2.57 \mu\text{M}$ (**2** and **4**) for 1 mg/ml, $16.1 \mu\text{M}$ (**1** and **3**) and $12.9 \mu\text{M}$ (**2** and **4**) for 5 mg/ml, $32.2 \mu\text{M}$ (**1** and **3**) and $25.7 \mu\text{M}$ (**2** and **4**) for 10 mg/ml.

^c Natural concentration of **1** to **4** in *Agelas* species. Sponge volume was determined by displacement of water with frozen material according to Pawlik *et al.* (1995).

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